

Synthesis of 2D and 3D Gold Nanoarrays and Semiconductor Nanoparticles in Supercritical Fluid Carbon Dioxide

Carlos A. Fernandez and Chien M. Wai

Department of Chemistry, University of Idaho, Moscow, Idaho 83844

Abstract

Metal and semiconductor nanoparticles of controllable size can be synthesized using microemulsion as template in different solvents. Two types of microemulsions have been studied using bis(2-ethylhexyl)sulfosuccinate (AOT) and a fluorinated AOT as surfactant for making water-in-oil and water-in-CO₂ microemulsions, respectively. Chemical reduction of metal ions (e.g. Au³⁺) dissolved in the water core of a microemulsion leads to the formation of metal nanoparticles (e.g. Au⁰) with size controllable by the dimension of the water core and the solvent surrounding the microemulsion. By mixing two microemulsions one containing Cd²⁺ and the other containing S²⁻, CdS nanoparticles of different sizes can be synthesized using the microemulsion-template method. Binary CdS/ZnS nanoparticles can also be synthesized using this method. Because the solvation strength of supercritical fluid CO₂ is tunable by varying temperature and pressure, we are developing techniques for synthesizing metal and semiconductor nanoparticles of controllable sizes using water-in-CO₂ microemulsions as template. The synthesized nanoparticles can be stabilized using an alkanethiol compound. Self-assembling of thiol-stabilized gold nanoparticles has been shown to result in ordered 2D and 3D arrays with sizes in micrometer dimensions. Characterization of these nanoparticles and nanoarrays are currently in progress.